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Privacy Policy iMouse™ is a registered trademark of Molex and used here without any implied or express warranty either express or implied. Molex does not claim ownership or any other right in the interface nor any third party process. Low-temperature infrared spectroscopic and theoretical studies of the adenine monoanion. The low-temperature infrared spectra of the ground electronic (lowest-energy) state of the adenine anion, Ad(-), are reported at 11.2 and 13.2 K in combination with quantum-chemical calculations. The experimental data are analyzed and compared to previous work on the synphylogenetic adenine cation, Ad(+). The structural assignments of the vibrational modes and the determination of the relative contributions of the out-of-plane bending modes of the purine ring, have been achieved using a quantum-chemical method with symmetry adapted perturbation theory. For both, the synphylogenetic and the ground electronic states of the anion, a similar pattern was found: the strong coupling of the NH<sub>2</sub> stretching vibration with the NH bending one of the purine ring leads to very intense inter-mode coupling patterns. The observed effects are rationalized in terms of a dynamical reorganization of the intermolecular electronic density. Theoretical calculations for the synphylogenetic and ground state of the anion indicate that the excitation of the NH<sub>2</sub> stretching vibration is a very efficient initial step for the dissociation process.

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